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PCT

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INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<p>(51) International Patent Classification <sup>6</sup> : <b>C09K 7/06</b></p>	<p><b>A1</b></p>	<p>(11) International Publication Number: <b>WO 95/06694</b> (43) International Publication Date: <b>9 March 1995 (09.03.95)</b></p>
<p>(21) International Application Number: <b>PCT/GB94/01877</b> (22) International Filing Date: <b>30 August 1994 (30.08.94)</b> (30) Priority Data: 9318099.0 1 September 1993 (01.09.93) GB 9413074.7 29 June 1994 (29.06.94) GB (71) Applicant (for all designated States except CA FR US): SOFITECH N.V. [BE/BE]; 5th floor, Atlantic House, Noorderlaan 147, Bus 5 C, B-2030 Antwerpen (BE). (71) Applicant (for CA only): SCHLUMBERGER CANADA LIMITED [CA/CA]; 24th floor, Monenco Place, 801 6th Avenue, S.W., Calgary, Alberta T2P 3W2 (CA). (71) Applicant (for FR only): COMPAGNIE DES SERVICES DOWELL SCHLUMBERGER, S.A. [FR/FR]; 50, avenue Jean-Jaurès, F-92541 Montrouge (FR). (72) Inventor; and (75) Inventor/Applicant (for US only): SAWDON, Christopher, Alan [GB/GB]; 6 Southview Road, Biscovey, Par, Cornwall PL24 2HJ (GB).</p>		<p>(74) Agent: NASH, David, Allan; Haseltine Lake &amp; Co., Hazlitt House, 28 Southampton Buildings, Chancery Lane, London WC2A 1AT (GB). (81) Designated States: AM, AT, AU, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, ES, FI, GB, GE, HU, JP, KE, KG, KP, KR, KZ, LK, LT, LU, LV, MD, MG, MN, MW, NL, NO, NZ, PL, PT, RO, RU, SD, SE, SI, SK, TJ, TT, UA, US, UZ, VN, European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG), ARIPO patent (KE, MW, SD).  Published <i>With international search report.</i></p>
<p>(54) Title: <b>WELLBORE FLUID</b> (57) Abstract  There is disclosed a biodegradable wellbore fluid having a continuous oil phase which comprises at least 50 % by weight of an n-alkane having from 10 to 20 carbon atoms, or a mixture of such alkanes.</p>		

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WELLBORE FLUID

This invention concerns wellbore fluids suitable for use in the oil and gas exploration and production industries and embraces fluids used for drilling, completion, workover or packing of wellbores. The invention is especially concerned with biodegradable wellbore fluids the liquid phase of which consists of an oil, or of water-in-oil "invert" emulsions, where the oil phase comprises n-alkanes having from 10 to 20 carbon atoms.

A wellbore is drilled by attaching to the bottom of a length of drill pipe a cutting bit which is rotated either by means of a prime mover on the surface which applies rotary motion to the top end of the drill pipe or by means of a turbine motor fixed to the drill pipe near the lower end, the turbine motor being driven by circulating fluid in the wellbore.

In either case a drilling fluid is required to remove cuttings from around the bit and to transport the cuttings to the surface for separation and removal. The drilling fluid also cools the bit and lubricates both bit and drill pipe thus reducing friction.

In many cases it is possible to use as the drilling fluid a simple mixture comprising water and a smectite clay which increases the viscosity of the fluid for the purposes of suspending and transporting the cuttings. However it is often necessary to drill a wellbore through geological formations the constituent materials of which swell and disintegrate on contact with water. In such cases an oil-based drilling fluid is used, that is a fluid in which the liquid phase consists of oil, or of water-in-oil "invert" emulsions.

Such oil based drilling fluids avoid the hydration problems caused by water contacting shale or salt formations. They also possess advantages compared to water based fluids of enhanced lubrication of the drill

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example in amounts up to about 50% by volume of the oil phase. In preferred embodiments of the invention, any polyalphaolefin is present in an amount no greater than 20% by volume, more preferably no greater than 10% by volume, and most preferably no greater than 5% by volume.

The continuous oil phase of the wellbore fluid of the present invention may contain up to 50% by weight of another biodegradable oil such as an ester oil as described in, for instance, EP-A-0374671 or EP-A-0374672 or a vegetable oil.

The drilling fluid of the invention has high biodegradability, both under aerobic and anaerobic conditions, and low toxicity.

The n-alkanes may be manufactured by the processing of renewable feedstocks obtained from vegetable oils or animal fats. For example, blends of n-alkanes, substantially free of other hydrocarbon types, and of suitable carbon chain length, may be produced directly from triglyceride oils and fats such as coconut oil or palm kernel oil. The vegetable oil can be reduced by hydrogen at high temperature and pressure (approximately 360°C and at 40 to 100 bar), in the presence of a catalyst, to yield substantially pure n-alkanes of carbon number corresponding to the fatty acids combined in the triglyceride ester feedstock. Propane is a by-product from the reduction of the glycerol component. The propane is readily separable from the n-alkanes because of its volatility. Small quantities of n-octane which may result can be undesirable with respect to the flash point of the product and may readily be removed by distillation if required, leaving a more preferable mixture of n-alkanes of C<sub>10</sub> - C<sub>18</sub> chain length.

Similarly, natural fatty acids may be decarboxylated or reduced to produce suitable n-alkanes

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of high purity. Fatty alcohols and alphaolefins may also be reduced to n-alkanes.

In a preferred embodiment of the present invention, the n-alkanes of the present invention have  
5 from 12 to 18 carbon atoms.

Surprisingly it has been found that the n-alkane mixtures of the present invention are highly advantageous for use as the continuous liquid phase of biodegradable oil based or invert emulsion wellbore  
10 fluids. (Invert emulsion wellbore fluids normally have a dispersed brine phase emulsified into the oil.)

The wellbore fluids of the present invention may contain added pour-point depressants to reduce the pour point (freezing point) of the n-alkane mixture when it  
15 is desired to utilise the drilling fluid in a cold environment.

The oil-continuous wellbore fluids may be used at oil water ratios preferably in the range 25 : 75 and 100 : 0.

20 The liquid components of the wellbore fluid according to the invention may comprise an emulsion of a dispersed hydrophilic liquid in the continuous oil phase.

In order to accelerate biodegradation, the  
25 wellbore fluid according to the present invention may further include added nutrients. Suitable nutrients include sources of the elements nitrogen, phosphorus and potassium, and preferably also trace elements such as iron required by bacteria. The nutrient may, for  
30 example, be dissolved in the aqueous phase if the wellbore fluid is an invert mud, or dispersed in the oil phase.

The wellbore fluid may also include an electron acceptor, for example, hematite, nitrate salts, and  
35 sulphate salts.

The principle of adding nutrients and/or an

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electron acceptor to a wellbore fluid may be extended to any wellbore fluid having a water-immiscible, or oily phase. Thus, in accordance with a further aspect of the present invention, there is provided a wellbore fluid comprising an oily phase characterised in that the wellbore fluid further comprises sources of one or more of the elements nitrogen, phosphorus and potassium, added as nutrients for the purpose of enhancing biodegradation of the said oil phase of the wellbore fluid. Preferably, the wellbore fluid comprises sources of each of the three elements.

In this aspect of the invention, the oil of the oil phase should be one which is capable of biodegrading in the presence of said nutrients under anaerobic conditions.

In the wellbore fluid of this further aspect of the invention, the wellbore fluid may additionally comprise sources of trace elements required by anaerobic bacteria for effective biodegradation and may additionally comprise an electron acceptor to aid in biodegradation.

Optional additives which may be contained in the wellbore fluid include: emulsifiers and other surface active agents, viscosifiers such as organoclays and polymers, filtration control agents such as Gilsonite<sup>(TM)</sup> and organically modified lignite, density increasing agents such as powdered barytes or hematite or calcium carbonate, or other wellbore fluid additives known to those skilled in the art.

Other oil miscible and biodegradable liquids such as oil soluble esters, vegetable oils, may be included as a minor proportion (e.g. less than 50% by weight) of the continuous oil phase of the wellbore fluid.

The emulsified water phase of the wellbore fluid may contain dissolved salts such as alkali metal halides. For example, sodium chloride, alkaline earth

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metal halides such as calcium chloride, or other water soluble organic compounds or salts for the purpose of adjusting the water activity of the dispersed aqueous phase (to enhance drilling performance).

5 Prior to the present invention it was not appreciated that biodegradable wellbore fluids could be formulated using n-alkanes as the continuous liquid phase, nor was it appreciated that drilling fluids could be formulated whose continuous phase comprised

10 substantially pure n-alkane mixtures with the advantageous properties of low plastic viscosity (as a consequence of the low kinematic viscosity of the n-alkane mixture), low enough pour point for all but the coldest environmental conditions, acceptable flash

15 point, minimal effect on human skin and minimal carcinogenicity, complete freedom from U.V. fluorescence, and minimal swelling effect on rubber goods.

According to a second aspect of the present

20 invention, there is provided a method of drilling a well wherein the drilling fluid is a wellbore fluid according to the first aspect of this invention. The cuttings which are contaminated with the drilling fluid may be discharged to the ocean floor or land farmed by

25 spreading the cuttings on the land whereupon natural biodegradation processes effectively remove the oil contamination.

According to a third aspect of the present invention, there is provided the use, as the continuous

30 oil phase of a wellbore fluid, of an n-alkane having from 10 to 20 carbon atoms.

The following examples illustrate the environmental and technical properties and performance of the wellbore fluids based on n-alkanes of the

35 present invention.

The wellbore fluid properties were tested in



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accordance with API RP 13B-2 1990.

The following abbreviations are used:

	KV	The kinematic viscosity of an oil in centistokes.
5	PV	The plastic viscosity of a drilling fluid (centipoise). Generally, a low PV is advantageous (as are low KV oils).
10	AV	The apparent viscosity of a drilling fluid (centipoise).
	YP	The yield point (lbs/100 ft <sup>2</sup> ) of the fluid and is a measure of the non Newtonian viscous characteristics.
15	6 rpm and 3 rpm	Dial readings on the Fann Viscometer which indicate the viscosity at low shear rates. Higher 6 rpm and 3 rpm values indicate greater thixotropy which is generally advantageous.
20	Gels	A measure of the gelling and suspending characteristics of the fluid (lbs/100 ft <sup>2</sup> ), determined using the Fann viscometer.
	ES	Electrical Stability of an emulsion (volts). high numbers imply high stability.
25	HTHP FL	High Temperature and High Pressure Fluid Loss. A measure of the ease of filtering a drilling fluid through a filter paper under HTHP conditions.
30		Results in millilitres of filtrate. Low filtrate volumes are advantageous.

#### EXAMPLE 1

Some properties of n-alkanes and other conventional or more recent base oils are compared in Table 1 which appears at the end of this description. In the table, base oil "Mixture C<sub>12,14,16,18</sub>" contained

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the following n-alkane approximate concentrations:

	<u>% Weight</u>
n-dodecane	52
n-tetradecane	19.5
5 n-hexadecane	10.5
n-octadecane	18

The above analysis is typical of the product obtained when coconut oil or palm kernel oil feedstocks are processed to yield n-alkanes, and the minor quantities of n-octane and n-decane are removed by distillation.

From Table 1 it is clear that C<sub>12</sub> to C<sub>18</sub> n-alkane blends possess advantageously low kinematic viscosity in comparison to linear alkylbenzene (LAB) and, in particular, Finagreen BDMF ester oil.

The n-alkane blend also compares favourably to LAB in terms of lower swelling effects on rubber goods and in the absence of fluorescence under ultra violet light. The hydrolytic stability of n-alkane blends is excellent, in contrast to ester oils which hydrolyse uncontrollably at temperatures in excess of circa 140°C.

#### EXAMPLE 2

The n-alkane (C<sub>12,14,16,18</sub>) mixture detailed in Example 1 was tested for its pour point by the Institute of Petroleum method IP15/67.

A similar test was performed on a sample of the n-alkane mixture which was dosed with 0.49% by weight of a commercial pour point depressant - EMPICRYL PPT 148\*.

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\*Trademark of Albright and Wilson Limited.

	Pour Point °C
(1) C <sub>12,14,16,18</sub> n-alkane mixture.	-2°C
(2) As (1) + 0.49% EMPICRYL PPT 148.	-9°C

- 5        The results show that the pour point of n-alkane mixtures may readily be reduced by the addition of a pour point depressant to allow effective use of the fluid under cold environmental conditions.

Table 1

Oil Type	KV cSt 22°C	KV Cst 40°C	Flash Point °C	Pour Point °C	Rubber Compatibility	U.V. Fluorescence	Hydrolytic Stability at above 140°C
n-DECANE	1.0	-	46	-29.7	Good	None	Good
n-DODECANE	1.5	-	74	-9.6	Good	None	Good
n-TETRADECANE	2.8	-	99	5.9	Good	None	Good
n-HEXADECANE	4.6	-	135	18	Good	None	Good
n-OCTADECANE	-	3.6	-	28	Good	None	Good
MIXTURE C <sub>12</sub> , 14, 16, 18 n-alkanes	-	2.0	91	-2	Good	None	Good
FINAGREEN BDMF*	-	5.9	>150	-23	Good	Moderate	Poor
(ESTER OIL)							
LINEAR ALKYL BENZENE		4.0		<-70	Moderate - Poor	Moderate	Good
(C <sub>10</sub> - C <sub>12</sub> )							
POLYALPHAOLEFIN		3.9 - 9.6	156 - 172	<-65	Good	None	Good
(M.I. DRILLING FLUIDS)							
CLAIRSOL 350 MHF*		1.95	93	-20	Good	Moderate	Good
MENTOR 28*		4.2	120	-15	Fair	Moderate	Good

\* Tradename

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## EXAMPLE 3

The n-alkane mixture of C<sub>12</sub> to C<sub>18</sub> paraffins described in Example 1 (hereinafter "NAM 1218"), was used as the continuous phase of a wellbore fluid of the following formulation.

Formulation per 350 mls of wellbore fluid

	NAM 1218	125 g
	INTERDRILL EMUL HT*	14g
	INTERDRILL OW*	1.5 g
10	TRUFLO 100*	2.5 g
	TRUDRILL S*	2.5 g
	TRUVIS HT*	3.5 g
	Lime	10 g
	OCMA ball clay	10 g
15	Calcium Chloride (86.5%)	18 g
	Water	43 g
	Barite	460 g

The properties of this wellbore fluid were compared to those of a similar formulation substituting a polyalphaolefin (essentially di-decene) for the NAM 1218.

Both fluids have a density of about 1.96 (16.3 lb/gal) and an oil:water ratio of 80:20.

The properties were tested both before (BHR) and after (AHR) heat ageing for 16 hours at 184°C. The viscous properties were measured at 50°C, and the HTHP FL at 178°C and 500 psi. The results obtained are illustrated in Table 2 below

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\*Mark of Schlumberger Dowell Limited

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Table 2

Property	Oil Phase			
	NAM 1218		Polyalphaolefin (PAO)	
	BHR	AHR	BHR	AHR
AV	51	54	105	125
PV	38	45	83	111
YP	26	18	44	28
6 rpm/3 rpm	13/11	9/8	20/18	11/9
Gels	13/20	11/18	20/32	11/20
ES	1055	980	1104	916
HTHP FL	-	17.6	-	16.4

10

The Apparent and Plastic viscosity (AV & PV) of the PAO fluid are more than double that of the NAM 1218 fluid as a consequence of the higher kinematic viscosity of the PAO. Low plastic viscosity is a definite advantage in minimising pressure drops on pumping and in minimising the "equivalent circulating density" (or the pressure exerted on possibly fragile downhole rock formations).

From the results after heat ageing (AHR), the NAM 1218 fluid provides good 6 rpm/3 rpm values and gel strengths (low shear rate viscosity and suspending ability), without paying the penalty of high plastic viscosity. The relatively small change in properties after heat ageing the NAM 1218 fluid shows the stability of the fluid at very high temperatures.

In contrast, emulsion fluids formulation using an ester oil continuous phase will undergo irreversible breakdown and thickening caused by hydrolysis of the ester, when exposed to such high temperature.

#### EXAMPLE 4

The n-alkane mixture of C<sub>12</sub> to C<sub>18</sub> paraffins described in Example 1 ("NAM 1218"), was used as the

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continuous phase of a wellbore fluid of the following formulation:

Formulations per 350 mls of wellbore fluid

	NAM 1218	167.4 g
5	Emulsifier	11.5 g
	Organoclay viscosifier	4 g
	Filtration reducer	6 g
	Lime	5 g
	Water	57.8 g
10	CaCl <sub>2</sub> (86.5% purity)	19.7 g
	OCMA ball clay	25.0 g
	Barite	121.7 g

The fluid exhibits a density of 1.2 (10 lb/gal) and an oil:water ratio of 80:20.

15 A fluid of similar formulation was also mixed using linear alkyl benzene(LAB) as the continuous phase.

Both fluids were tested for biodegradability under aerobic conditions by the method OECD 301F. The results obtained are illustrated in Table 3 below.

Table 3

Fluid Type	Biodegradability %
NAM 1218	100%
LAB	66%

25

The NAM 1218 based wellbore fluid is of exceptionally good biodegradability under aerobic conditions.

#### EXAMPLE 5

30 The NAM 1218 wellbore fluid formulation as in Example 4 was compared for anaerobic biodegradability against similar formulations based upon mineral oil, linear alkyl benzene and 2-ethylhexyl oleate. The anaerobic biodegradation test method was ECETOC 28 (56 day test).

35

A second NAM 1218 formulation included Ammonium

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Sulphate (3.1g) and potassium dihydrogen phosphate (2.4g) mixed into the brine phase to ascertain whether such nutrient addition would accelerate the biodegradation process. The results obtained are illustrated in Table 4 below.

Table 4

Base Fluid Type	Anaerobic Biodegradability %
Mineral Oil	0
Linear alkyl benzene	0
NAM 1218	10.2
NAM 1218 + nutrients	24.9
2-ethylhexyl oleate	48.3

10

15

Anaerobic biodegradation of predominantly hydrocarbon materials to carbon dioxide and methane much less energetically favourable than biodegradation under aerobic conditions to carbon dioxide. Hence lower results, and slower rates, are normal.

20

As expected the mineral oil, which contains much branched and cyclic hydrocarbons, did not biodegrade. Neither did the linear alkyl benzene based drilling fluid.

25

However the NAM 1218 fluid displayed significant anaerobic biodegradation at 10.2%, which is catalysed to 24.9% by the addition of nutrient salts providing the elements N, P and K and an electron acceptor as sulphate. These are surprisingly good results.

30

For comparison, the ester fluid (2-ethylhexyl oleate) is more biodegraded at 48.3%, but such fluids cannot be used for hot well conditions above about 140°C because of hydrolysis of the ester.

#### EXAMPLE 6

35

The NAM 1218 wellbore fluid formulation as in



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Example 4 was tested for toxicity to *Corophium volutator*, a seabed sediment dweller. Similar formulations using alternative oil phases were compared as shown in Table 5.

5

Table 5

Fluid Type	Toxicity to <i>Corophium</i>
	LC <sub>50</sub> (mg/kg sediment)
Mineral Oil	294
Linear Alkyl Benzene	210
NAM 1218	1654

10

The results show the very advantageous increase in LC<sub>50</sub> (reduction in toxicity) provided by the NAM 1218 wellbore fluid.

CLAIMS

1. A biodegradable wellbore fluid having a continuous oil phase which comprises at least 50% by weight of an n-alkane having from 10 to 20 carbon atoms, or a mixture of such alkanes.
2. Wellbore fluid according to Claim 1, wherein the n-alkanes have from 12 to 18 carbon atoms.
3. Wellbore fluid according to claim 1 or 2, wherein the n-alkanes are derived from the chemical processing of natural fats, vegetable oils, naturally derived fatty acids, or naturally derived fatty alcohols.
4. Wellbore fluid according to any preceding claim, further comprising an effective amount of a pour point depressant.
5. Wellbore fluid according to any preceding claim, whose liquid components comprise an emulsion of a dispersed hydrophilic liquid in the continuous oil phase.
6. Wellbore fluid according to any of the preceding claims, further including added nutrients to accelerate biodegradation.
7. Wellbore fluid according to any of the preceding claims, further including an electron acceptor to accelerate anaerobic biodegradation.
8. Wellbore fluid according to claim 7, wherein the electron acceptor is chosen from hematite, nitrate salts, and sulphate salts.
9. Wellbore fluid according to any preceding claim, wherein the oil phase further includes a biodegradable oil-miscible component.
10. A wellbore fluid according to any preceding claim, which is substantially free of aromatic hydrocarbons, cycloparaffins, cycloalkenes iso-alkanes.
11. A wellbore fluid according to claim 10 which is additionally substantially free of polyalphaolefins.

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12. A wellbore fluid according to any one of the preceding claims which contains no more than 35% by weight of branched chain alk-1-enes.

13. A method of drilling a well wherein the  
5 drilling fluid is a wellbore fluid according to any preceding claim.

14. The use, as the continuous oil phase of a wellbore fluid, of an oil comprising an n-alkane having from 10 to 20 carbon atoms, or mixtures thereof.

# INTERNATIONAL SEARCH REPORT

International Application No

PCT/GB 94/01877

A. CLASSIFICATION OF SUBJECT MATTER  
IPC 6 C09K7/06

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C09K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US,A,4 508 628 (T.O.WALKER) 2 April 1985 see column 3, line 59 - column 4, line 57 see example 1 ---	1,5,10, 11,13,14
Y	US,A,4 012 329 (J.B.HAYES) 15 March 1977 see column 3, line 57 - column 4, line 9 see column 5, line 24 - line 35 ---	1,5,10, 11,13,14
Y	EP,A,0 134 173 (RHONE-POULENC SPECIALITES) 13 March 1985 see page 1, line 1 - line 7 see page 3, line 28 - line 33 see page 5, line 17 - line 26; examples 1,2 --- -/--	1,5,10, 11,13,14

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

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Date of the actual completion of the international search

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# INTERNATIONAL SEARCH REPORT

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## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	<p>WO,A,83 02949 (R.JACHNIK) 1 September 1983</p> <p>see page 7, line 3 - page 8, line 24  see page 9  see page 10, line 5 - page 11, line 24  see claims 1,5,10</p> <p>---</p>	<p>1,4,5,  9-11,13,  14</p>
Y	<p>GB,A,2 212 192 (EXXON CHEMICAL PATENTS INC.) 19 July 1989</p> <p>see page 3, line 9 - line 30  see page 9, line 2 - line 16; claims 1,4,9</p> <p>---</p>	<p>1,5,  9-11,13,  14</p>
Y,P	<p>US,A,5 333 698 (D.C.VAN SLYKE) 2 August 1994</p> <p>see column 2, line 10 - column 3, line 16  see column 5, line 7 - line 37  see column 7, line 40 - line 63</p> <p>-----</p>	<p>1,4,5,  10,13,14</p>

# INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

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Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US-A-4508628	02-04-85	NONE	
US-A-4012329	15-03-77	NONE	
EP-A-0134173	13-03-85	FR-A- 2550796 OA-A- 7795	22-02-85 20-11-86
WO-A-8302949	01-09-83	AU-A- 1166683 FR-A- 2521583 GB-A- 2115458 NL-A- 8300609 AU-A- 1166483 AU-A- 1166583 FR-A- 2521582 FR-A- 2521584 WO-A- 8302950 WO-A- 8302951 GB-A- 2117429 GB-A- 2115459 NL-A- 8300607 NL-A- 8300608	25-08-83 19-08-83 07-09-83 16-09-83 25-08-83 25-08-83 19-08-83 19-08-83 01-09-83 01-09-83 12-10-83 07-09-83 16-09-83 16-09-83
GB-A-2212192	19-07-89	NONE	
US-A-5333698	02-08-94	NONE	